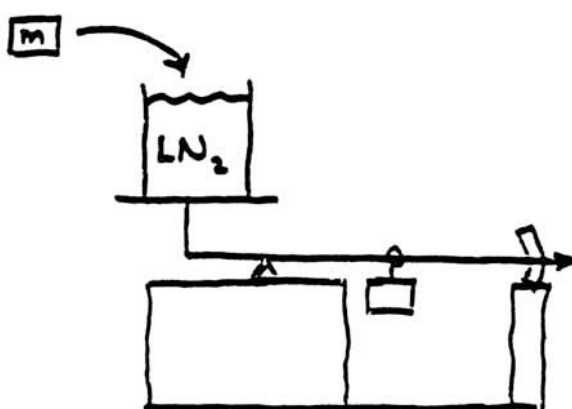


Experimental Problem 1 -- Solutions

Method #1



$$Q = mc\Delta T = m \int c dT$$

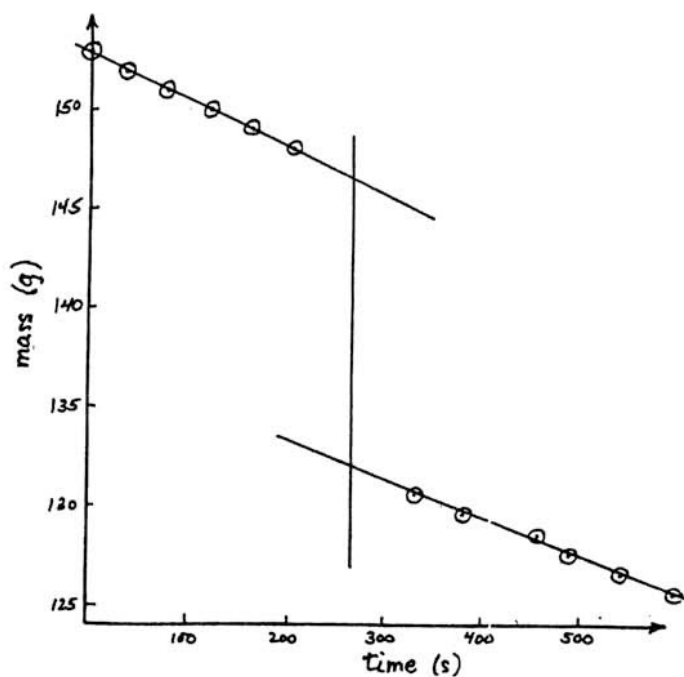
$$Q = L \Delta M_{LN_2}$$

$$m = 19.4 \pm 0.1 \text{ g}$$

total mass	clock time	time
153 g	0:00.0	0
152	0:36.8	36.8
151	1:19.1	79.1
150	2:00.7	120.7
149	2:40.5	160.5
148	3:23.1	203.1

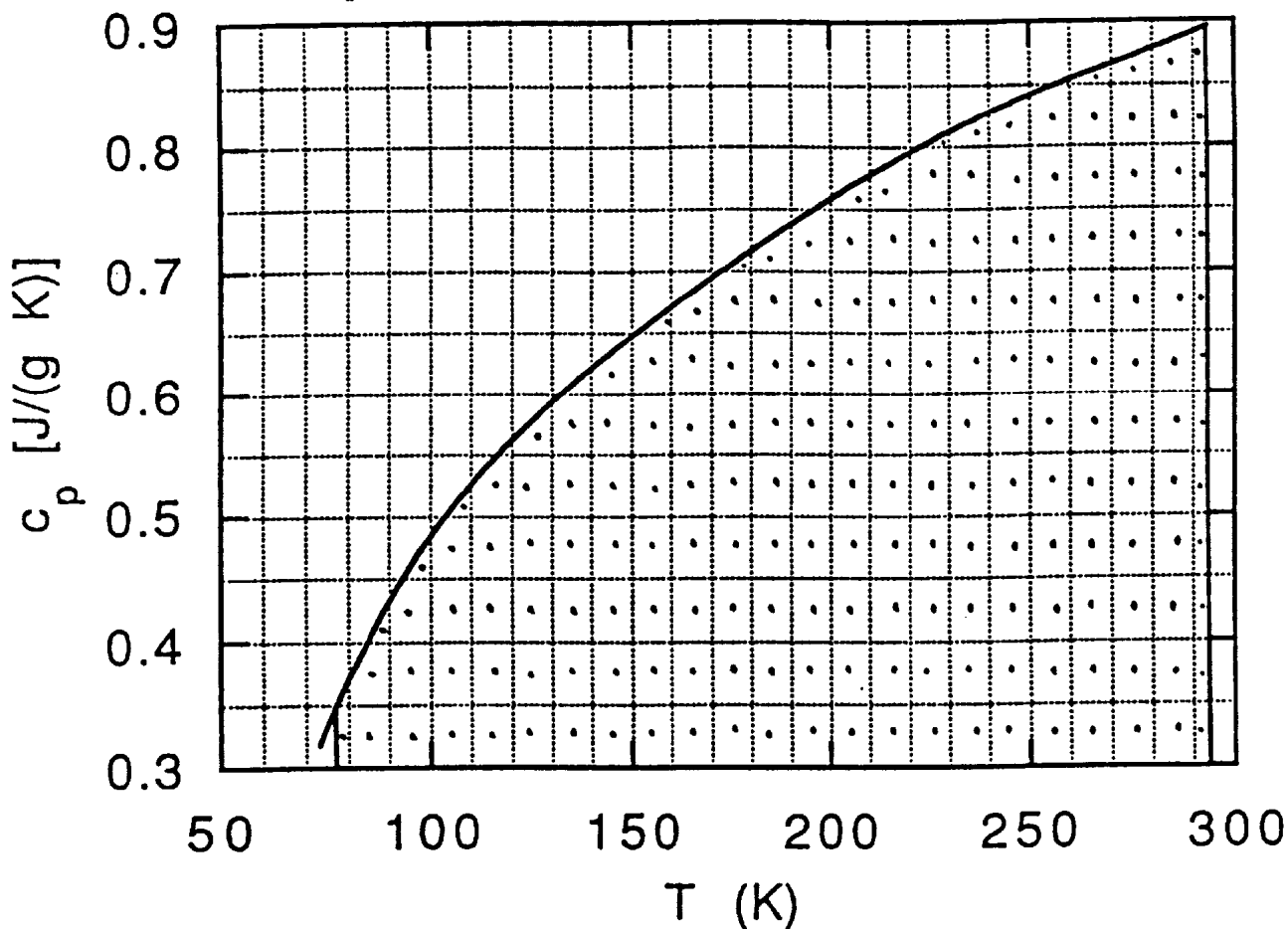
Add Al mass

150 (130.6)	5:31.8	331.8
149 (129.6)	6:21.6	381.6
148 (128.6)	7:17.3	457.3
147 (127.6)	8:08.6	488.6
146 (126.6)	9:00.9	540.9
145 (125.6)	9:54.6	594.6



$$\begin{aligned} \Delta M_{LN_2} &= 146.5 - 132.0 \\ &= 14.5 \pm 0.3 \text{ g} \end{aligned}$$

Specific Heat of Aluminum



$$\int_{77}^{293} c \, dT \approx (0.3)(293 - 77) + (173)(0.5)$$

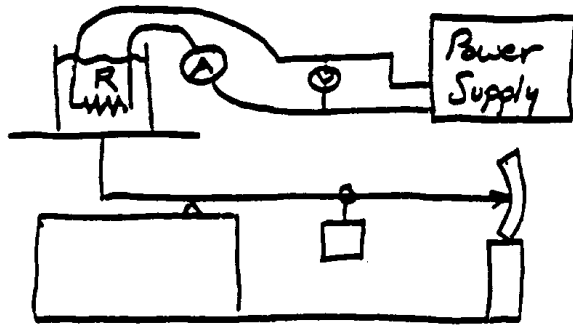
$$\approx 64.8 + 86.5 = 151 \pm 2 \text{ J/g}$$

$$Q = \int m c \, dT = (19.4 \pm 0.1 \text{ g})(151 \pm 2 \text{ J/g})$$

$$= 2930 \pm 42 \text{ J.}$$

$$L = \frac{Q}{\Delta M_{\text{LN}_2}} = \frac{2930 \pm 42 \text{ J}}{14.5 \pm 0.3 \text{ g}} = 202 \pm 5 \text{ J/g}$$

Method #2



$$P = IV = V^2/R = I^2R$$

$$P = \Delta Q/\Delta t$$

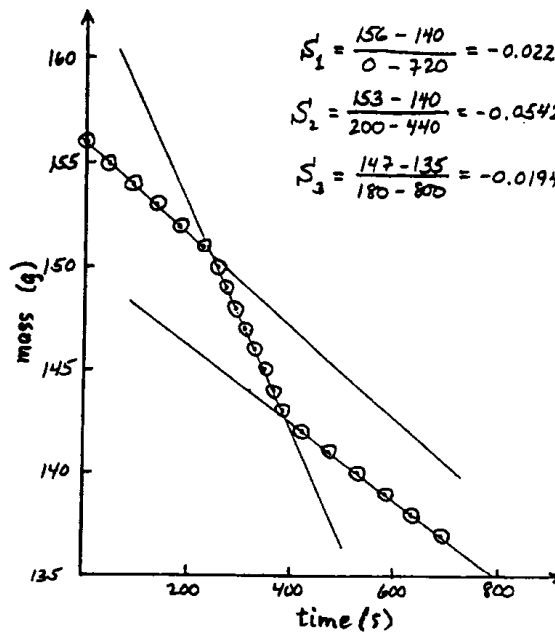
$$Q = M_{LN_2}L$$

$$R = 23.0 \Omega \text{ (in LN}_2\text{)}$$

$$V = 12.7 \text{ V}$$

$$I = 0.56 \text{ A}$$

	total mass	clock time	time
$P=0$	156 g	0:00.0	0 s
	155	0:45.2	45.2
	154	1:31.4	91.4
	153	2:16.2	136.2
	152	2:60.0	180.0
	151	3:47.2	227.2
$P \neq 0$	150	4:13.6	253.6
	149	4:32.1	272.1
	148	4:50.1	290.1
	147	5:08.9	308.9
	146	5:27.2	327.2
	145	5:45.7	345.7
	144	6:04.1	364.1
	143	6:21.9	381.9
$P=0$	142	7:02.3	422.3
	141	7:58.4	478.4
	140	8:51.2	531.2
	139	9:43.7	583.7
	138	10:34.6	634.6
	137	11:30.7	690.7



$$S_{P \neq 0} = -0.054 \pm 0.001 \text{ g/s}$$

$$\langle S_{P=0} \rangle = -0.020 \pm 0.001 \text{ g/s}$$

$$\text{Power} = P = \left| \frac{Q}{\Delta t} \right| = L \left| \frac{\Delta M_{LN_2}}{\Delta t} \right|$$

$$\left. \begin{aligned} P = IV &= 7.11 \text{ W} \\ P = I^2R &= 7.21 \text{ W} \\ P = V^2/R &= 7.01 \text{ W} \end{aligned} \right\} P = 7.1 \pm 0.1 \text{ W}$$

$$|\Delta M_{LN_2}/\Delta t| = 0.054 - 0.020 = 0.034 \pm 0.0014 \text{ J/s}$$

$$L = \frac{P}{\Delta M_{LN_2}/\Delta t} = \frac{7.1 \pm 0.1}{0.034 \pm 0.0014} = 209 \pm 9 \text{ J/g}$$

Experimental Problem 1: Grading Scheme

Method No. 1 (5 points maximum)

- 1) 0.5 Uses $Q = mc\Delta T$ or $Q = m \int c dT$
- 2) 0.5 Uses $Q = L\Delta M_{\text{LN}_2}$
- 3) 0.5 Measures mass of aluminum correctly
- 4) 0.5 Measures ΔM_{LN_2} in some way
- 5) 0.5 Takes into account "thermal leakage" in some way and corrects for aluminum added to container
- 6) 0.5 Takes into account "thermal leakage" not being constant in time
- 7) 0.5 Uses reasonable values for c and ΔT or does $\int c dT$ integral in a reasonable way
- 8) 0.5 No mistakes made in computing L
- 9) 0.5 Error estimate is reasonable for methods used
- 10) 0.5 Value for L is within bounds set by grading team using good procedures

Method No. 2 (5 points maximum)

- 1) 0.5 Uses $P = \Delta Q/\Delta t$
- 2) 0.5 Uses $P = IV = I^2R = V^2/R$
- 3) 0.5 Uses $Q = LM_{\text{LN}_2}$
- 4) 0.5 Measures two parameters (to get P) correctly
- 5) 0.5 Measures M_{LN_2} in some way
- 6) 0.5 Takes into account "thermal leakage" in some way
- 7) 0.5 Takes into account "thermal leakage" not being constant in time
- 8) 0.5 No mistakes made in computing L
- 9) 0.5 Error estimate is reasonable for methods used
- 10) 0.5 Value for L is within bounds set by grading team using good procedures